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Solution: $p = z \rho g = 30 \times 1878 \times 9.65 \text{ Pa} = 543.681 \text{ kPa}$ Q1.7 Assume that the pressure p and the specific volume v of the atmosphere are related according to the equation $p v^{1.4} = 2.3 \times 10^5$, where p is in N/m² abs and v is in m³/kg. The acceleration due to gravity is constant at 9.81 m/s². What is the depth of atmosphere necessary to produce a pressure of

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Temperature Some Important Notes Comparison of Temperature scale Relation: C 0 100 0 – – = F 32 212 32 – – = K 273 373 273 – – = 0 80 0 p – – = – x 10 30 10 Questions with Solution P. K. Nag Q2.1 The limiting value of the ratio of the pressure of gas at the steam point and at the triple point of water when the gas is kept at constant volume is found to be 1.36605.

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