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CHAPTER 2 REVIEW Measurements and Calculations SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research team records the rainfall in inches

2 Measurements and Calculations

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CHAPTER 2 REVIEW Measurements and Calculations SECTION 2-3 SHORT ANSWER Answer the following questions in the space provided. 1. Report the number of significant figures in each of the following values: a. 0.002 37 g d. 64 mL b. 0.002 037 g e. 1.3 102 cm c. 350. J f. 1.30 102 cm 2. Write the value of the following operations using scientific notation:

2 Measurements and Calculations

1: Chemistry Is a Physical Science: Section 1 Review: p.5: 2: Matter and Its Properties: Section 2 Review: p.14: 3: Elements: Section 3 Review: p.20: Chapter Review: p.22

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

$2\text{N}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 4\text{NO}(\text{g})$ a. Balance the equation. $2\text{N}_2(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 4\text{NO}(\text{g}) + 2\text{H}_2\text{O}(\text{g})$ b. What is the mole ratio of NO_2 to O_2 ? 15.0 mol c. If 20.0 mol of NO_2 form, how many moles of O_2 must have been consumed? True d. Twice as many moles of NO_2 form as moles of N_2O are consumed. True or False? False e. Twice as many grams of NO_2 ...

mc06se cFMs r i-vi

$2\text{SO}_4^{2-}(\text{s}) + 3\text{Fe}^{3+}(\text{aq}) + 3\text{SO}_4^{2-}(\text{aq})$ 5. a. Write the net ionic equation for the reaction that occurs when solutions of lead(II) nitrate and ammonium sulfate are combined. $\text{Pb}^{2+}(\text{aq}) + \text{SO}_4^{2-}(\text{aq}) \rightarrow \text{PbSO}_4(\text{s})$ b. What are the spectator ions in this system? NO_3^- and NH_4^+ are spectator ions. MODERN CHEMISTRY IONS IN AQUEOUS SOLUTIONS AND ...

13 Ions in Aqueous Solutions and Colligative Properties

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(Section 2.4) Describe the relationship between stability and potential energy. (Section 7.1) Review Skills The presentation of information in this chapter assumes that you can already perform the tasks listed below. You can test your readiness to proceed by answering the Review Questions at the end of the chapter.

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