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(ii) Among CH 3 NH 2 and (C 2 H 5) 2 NH, primarily due to the greater +I-effect of the two -C 2 H 5 groups over one -CH 3 group, (C 2 H 5) 2 NH is more basic than CH 3 NH 2.In both C 6 H 5 NH2 and C 6 H 5 N(CH 3) 2 lone pair of electrons present on N-atom is delocalized over the benzene ring but C 6 H 5 N(CH 3) 2 is more basic due to +I effect of two-CH 3 groups.

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13 Ions in Aqueous Solutions and Colligative Properties

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Chapter 13 Properties of Solutions

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