

Basic Concepts Of Chemistry 9th Edition

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The Lewis dot symbol for an element represents the valence electrons as dots around the symbol of the element. The first four valence electrons are shown as single dots placed on four sides of the symbol, the subsequent electron are shown as dots paired up with first four dots.

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Kotz 9e Chapter 1 Basic Concepts of Chemistry 5. Which of the following statements concerning the kinetic-molecular theory of matter is/are CORRECT? 1. Particles in a gas move faster as the temperature increases. 2. Particles in a liquid are packed closely together, but are not confined to specific positions. 3.

Chapter 1 Basic Concepts of Chemistry

In simple terms: Protons and neutrons make up the inner part of the atom, or nucleus, and account for most of the mass of the atoms. Electrons make up a cloud (orbitals, or areas in which the probability of finding an electron is high) around the nucleus. Bohr model of an atom, via universesetoday.com.

15 Essential Basic Chemistry Concepts Explained

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Chemistry is the branch of science that deals with the study of the composition, structure, properties of matter and changes it undergoes. Anything that has mass and occupies space is known as matter. Matter can exist in three physical states based on the arrangement of constituent particles – solid, liquid and gaseous.

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